

Historical Development of the Early and current Atomic Model

Pre-AP

For the following individuals research what they contributed to the discovery and advancement of knowledge of what we consider the modern atom. Treat the activity as a note taking session, as you will need these later to study. You may type out the answers or write them in your notes.

Democritus (400 B.C.) –

What does the Greek word “atomos” mean?

What were his postulates regarding atoms?

Were his thoughts based on the scientific method, why or why not?

John Dalton (1808) –

What are Dalton's four main components of atomic theory?

What was his atomic theory based on?

How did atoms combine according to Dalton?

Julius Plucker (1859) –

What major discovery did Plucker make that later led to the identification of the electron?

How does this apparatus work?

Sketch out a schematic of its function?

G.J. Stoney (1879) – Stoney proposed that electricity was made of what?

Sir William Crookes (1879) – Crookes' experiment with a cathode ray led to what being learned about the rays:

1. Direction of travel of ray;
2. Glass does what in current;
3. Something to do with a charged particle;
4. Effect of magnet on field/ ray;
5. Effect on pinwheels showed rays had what?

Eugene Goldstein (1886) –

Goldstein used a cathode ray to find this type of subatomic particle?

How does the mass of this particle compare to the particle mass of an electron (need a number ratio)?

Roentgen (1895) -

While using a cathode ray tube Roentgen observed that nearby chemicals glowed, this was caused by what type of energy that is not affected by a magnetic field?

Becquerel (1896) – Becquerel made what discovery when he placed a uranium compound he had put into a drawer darkened a photographic film stored in the drawer?

He discovered some chemicals spontaneously _____ and give off _____ rays.

This discovery supported the idea of a _____ atom.

J.J. Thomson (1897-1904) - His experiment with a cathode ray tube led to the discovery of what particle (hint C/g)?

What model of the atom did Thomson propose, draw it out and describe it? How was this different from later models?

Ernest Rutherford (1898) –

Try to describe Rutherford's famous experiment? What were the major findings?

What did he win the Nobel Prize for?

What three particles are he credited with naming and characterizing?

Max Planck (1901) – Planck discovered that energy flowed how?

These were called?

What are the components of a quantum of light formula, what does each stand for?

Robert Millikan (1909) –

Millikan conducted what classic experiment?

What did this experiment determine?

Erwin Schroedinger (1926) –

Schroedinger used mathematics and statistics and other discoveries to define what student of physics/ chemistry?

James Chadwick (1932) –

Found what subatomic particle?

How was this found?